

Available online at www.sciencedirect.com



Tetrahedron Letters 45 (2004) 6963-6965

Tetrahedron Letters

# Rapid and facile Lewis acid catalysed Boc protection of amines $\stackrel{\approx}{\sim}$

G. V. M. Sharma,\* J. Janardhan Reddy, P. Sree Lakshmi and Palakodety Radha Krishna

D-211, Discovery Laboratory, Organic Chemistry Division-III, Indian Institute of Chemical Technology, Hyderabad 500 007, India

Received 28 April 2004; revised 7 July 2004; accepted 16 July 2004 Available online 7 August 2004

Abstract—Efficient Boc protection of amines was carried out using  $(Boc)_2O$  in the presence of a catalytic amount of  $ZrCl_4$  (10 mol%) in acetonitrile at room temperature. The reaction times are very short and the yields are generally high. © 2004 Elsevier Ltd. All rights reserved.

# 1. Introduction

Among the various protecting groups used for amines, Boc protection has become a fundamental tool of modern peptide synthesis<sup>1</sup> and particularly of the Merrifield strategy for solid-phase peptide synthesis.<sup>2</sup> Owing to the instability of the corresponding *t*-butyl chloroformate and the explosive properties of t-butylazido formate, the (Boc)<sub>2</sub>O reagent is widely used for the introduction of the *t*-butoxycarbonyl group. Even though a variety of base mediated reaction conditions are available for Boc protection, the only reported acid (Yttria-Zirconia) mediated reaction conditions<sup>3</sup> need longer reaction times (3-48 h). In pursuance of our work on new synthetic methods<sup>4</sup> and non-natural peptides,<sup>5</sup> we were interested in exploring the possibility of developing Lewis acid catalysed reaction conditions for Boc protection. Herein, we report the ZrCl<sub>4</sub> catalysed Boc protection of amines with short reaction times and high yields.

$$R-NH_2 \xrightarrow{(Boc)_2O, ZrCl_4(10 \text{ mol}\%)}_{CH_3CN, rt} R-NHBoc$$
(1)

Aniline (Table 1, entry 1) and  $(Boc)_2O$  in acetonitrile were treated with 10mol% ZrCl<sub>4</sub> at room temperature to afford **1a** (95%) in 3min. The same reaction with Yttria–Zirconia as catalyst<sup>3</sup> took 14h, while it required 48h in the absence of any catalyst. This interesting result prompted us to explore the reactivity of (Boc)<sub>2</sub>O with a variety of amines in the presence of ZrCl<sub>4</sub>. Accordingly, aryl/heteroarylalkyl (entries 2 and 3), cycloalkyl (entries 4 and 5), acetoxyalkyl amines (entry 8) and secondary amines (entries 9, 10 and 11) underwent smooth Boc protection to furnish the corresponding products (Boc protected amines) in good to excellent yields (Table 1) in 3–10min. It is worth mentioning that aminols (entries 6 and 7) on treatment with (Boc)<sub>2</sub>O in acetonitrile at room temperature chemoselectively gave **6a** and **7a** in high yields.

In a further study, amino acid esters (Table 2, entries 1, 2 and 3) were converted to the corresponding N-Boc esters under similar reaction conditions in 10min and in good yields. However, the methyl ester of histidine (Table 2, entry 5) on reaction with 1 mol of (Boc)<sub>2</sub>O gave 16a (82%) as the sole product in 10min, while with 2 mol it gave 16b (80%) in 15 min. The chemoselective protection of cysteine (Table 2, entry 4) gave the N-Boc protected derivative 15a (81%) in 10min. This procedure could conveniently be applied to a variety of amino acids. In a continued study, the C-linked carbo  $\beta$ -amino acid esters derived from sugars (Table 2, entries 6, 7 and 8) underwent smooth Boc protection in the presence of ZrCl<sub>4</sub> (10mol%) in 15–25min to give 17a, 18a and 19a, respectively, in high yields. The conventional procedure  $[(Boc)_2O, Et_3N, THF]$  for the same transformation took more than 2h.

In conclusion, this protocol is operationally simple, rapid and high yielding. The reaction conditions are mild and inexpensive involving the use of a readily available and environmentally friendly catalyst  $(ZrCl_4)$  at room temperature. The formation of side products was not observed.

*Keywords*: Amines; Amino acids; Carbo β-amino acids; Lewis acid catalyst; (Boc)<sub>2</sub>O.

<sup>&</sup>lt;sup>☆</sup>IICT Communication No. 040422.

<sup>\*</sup> Corresponding author. Fax: +91-(0)40-27160387; e-mail: esmvee@ iict.res.in

<sup>0040-4039/\$ -</sup> see front matter @ 2004 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2004.07.072

Entry	Starting material	Product	Time (min)	Yield (%)
	RNH <sub>2</sub>	RNHBoc		
1	R = phenyl 1	$\mathbf{R} = \mathbf{phenyl} \ \mathbf{1a}^{\mathbf{a}}$	3	95
2	R = benzyl 2	$\mathbf{R} = \text{ benzyl } \mathbf{2a}^{\mathrm{a}}$	3	96
3	R = 4-pyridylmethyl <b>3</b>	$\mathbf{R} = 4$ -pyridylmethyl $3\mathbf{a}^{c}$	3	90
4	R = cyclohexyl 4	$\mathbf{R} = \text{cyclohexyl } \mathbf{4a}^{a}$	3	91
5		N NHBoc	5	88
6	HO NH <sub>2</sub> 6	HO NHBoc 6a <sup>a</sup>	5	96
7	7  R' = H	NHBoc	5	90
8	$8 \mathbf{R}' = \mathbf{A} \mathbf{c}$	$7\mathbf{a}^{\mathrm{b}} \mathbf{R}' = \mathbf{H}$ $8\mathbf{a}^{\mathrm{c}} \mathbf{R}' = \mathbf{A}\mathbf{c}$	10	85
9		$\mathbb{N}^{\mathbb{N}}$	5	92
10	XNH	XNBoc	5	92
	<b>10</b> X = $CH_2$	$10a^{b} X = CH_{2}$		
11	11 X = O	$11a^{a} X = O$	10	90

Table 1. ZrCl<sub>4</sub> (10mol%) catalysed protection of amines

<sup>a</sup> See Ref. 3.

<sup>b</sup> Commercially available compounds.

<sup>c</sup> All new compounds gave satisfactory spectral and analytical data.

# 2. General experimental procedure

To a solution of the amine/amino acid ester (1 mmol) in acetonitrile (2 mL), (Boc)<sub>2</sub>O (1 mmol) was added dropwise followed by a catalytic amount of ZrCl<sub>4</sub> (10 mol%) at room temperature. After stirring the reaction mixture for the specified time (Tables 1 and 2), the solvent was removed under reduced pressure and the mixture diluted with water and extracted with ethyl acetate ( $3 \times 5$  mL). The combined organic layers were washed with water (10 mL), brine (10 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), evaporated under reduced pressure and purified the residue by column chromatography (silica gel, 60–120 mesh, 5–15% EtOAc in hexane) to furnish the pure product.

### 2.1. Spectral data of selected compounds

**2.1.1. 4-**(*N*-*tert*-**Butylcarboxylate**)-amino methyl pyridine (3a). White solid, mp 70–75 °C; IR (KBr): 3380, 2985, 2928, 1705, 1500, 1310, 1165, 1070, 1010 cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>):  $\delta$  1.40 (s, 9H), 4.32 (d, 2H, J = 6.6 Hz), 5.00 (br s, 1H), 7.16 (d, 2H, J = 5.3 Hz), 8.55 (d, 2H, J = 5.3 Hz); EIMS: (m/z, %) 208 (M<sup>+</sup>, 3), 152 (M<sup>+</sup>-56, 73), 108 (M<sup>+</sup>-Boc, 24), 57 (100). Anal. Calcd for C<sub>11</sub>H<sub>16</sub>N<sub>2</sub>O<sub>2</sub> (208): C, 63.44; H, 7.74. Found: C, 63.47; H, 7.70.

**2.1.2.** *tert*-Butyl 1*H*-1-imidazolecarboxylate (9a). White solid, mp 42–45 °C, lit. (Lancaster catalogue) mp 46–47 °C; IR (KBr): 3365, 2975, 2930, 1725, 1700, 1502, 1308, 1165, 1070,  $1015 \text{ cm}^{-1}$ ; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  8.02 (s, 1H), 7.36 (s, 1H), 7.00 (s, 1H), 1.62

(s, 9H, Boc); EIMS: (m/z, %) 169  $(M+1^+, 8)$ , 131  $(M+1^+-37, 9)$ , 96  $(M+1^+-72, 100)$ , 69  $(M+1^+-Boc, 63)$ .

**2.1.3.** Methyl (3*S*)-3-[(*tert*-butoxy)carbonylamino]-3-[6methoxy-2,2-dimethyl-(3*aR*,6*S*,6*aR*)-tetrahydrofuro-[2,3*d*][1,3]dioxol-5-yl]propanoate (17a).  $[\alpha]_D^{25}$  -26.9 (*c* 1.1, CHCl<sub>3</sub>); IR (neat): 3385, 2980, 2938, 1725, 1705, 1502, 1308, 1161, 1071, 1013 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  1.31 (s, 3H), 1.43 (s, 9H), 1.48 (s, 3H), 2.67 (dd, 1H, *J* = 7.9, 14.6Hz), 2.71 (dd, 1H, *J* = 3.2, 14.6Hz), 3.37 (s, 3H), 3.68 (s, 3H), 3.68 (d, 1H, *J* = 3.1Hz), 4.30 (m, 1H), 4.29 (m, 1H), 4.57 (d, 1H, *J* = 3.8Hz), 5.09 (br s, 1H), 5.91 (d, 1H, *J* = 3.8Hz); FABMS: (*m*/*z*, %) 752 (2[M+1]<sup>+</sup>, 4), 376 (M+1<sup>+</sup>, 23), 320 (12), 276 (M+1<sup>+</sup>-Boc, 100), 218 (11), 133 (15). Anal. Calcd for C<sub>17</sub>H<sub>29</sub>NO<sub>8</sub> (375): C, 54.39; H, 7.79. Found: C, 54.45; H, 7.72.

**2.1.4.** Methyl (3*S*)-3-[(*tert*-butoxy)carbonylamino]-3-[2,2,7,7-tetramethyl-(3*aR*,5*R*,5*aS*,8*aS*,8*bR*)-tetrahydrodi-[1,3]dioxolo[5,4-*b*:4,5-*d*]pyran-5-yl]propanoate (19a).  $[\alpha]_{D}^{25}$ -41.0 (*c* 1.0, CHCl<sub>3</sub>); IR (neat): 3380, 2985, 2940, 1725, 1710, 1500, 1308, 1165, 1070, 1012 cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  1.33 (s, 6H), 1.42 (s,12H), 1.48 (s, 3H), 2.75 (d, 2H, *J* = 7.7 Hz), 3.56 (s, 3H), 3.94-4.02 (m, 1H), 4.06 (q, 1H, *J* = 7.7 Hz), 4.24-4.28 (m, 2H), 4.55 (d, 1H, *J* = 18.4 Hz), 4.98 (br s, 1H), 5.48 (d, 1H, *J* = 6.4 Hz); FABMS: (*m*/*z*, %) 862 (2M<sup>+</sup>, 5), 432 (M+1<sup>+</sup>, 20), 376 (12), 332 (M+1<sup>+</sup>-Boc, 100), 274 (11), 189 (15), 64 (25). Anal. Calcd for C<sub>20</sub>H<sub>33</sub>NO<sub>9</sub> (431): C, 55.67; H, 7.71. Found: C, 55.59; H, 7.65.

Table 2. ZrCl<sub>4</sub> (10mol%) catalysed protection of amino acid esters

Entry	Starting material	Product	Time (min)	Yield (%)
1	$\frac{MeO_2C}{12} NH_2$	MeO <sub>2</sub> C NHBoc 12a <sup>a</sup>	10	82
2	$ \frac{\underline{NH}_{2}}{CO_{2}Me} $ 13	$ \underbrace{\overset{\mathrm{NHBoc}}{\leftarrow}}_{\mathrm{CO}_{2}\mathrm{Me}} $ 13a <sup>a</sup>	10	84
3	$\frac{\mathrm{NH}_{2}\mathrm{HCl}}{\mathrm{MeO}_{2}\mathrm{C}}$ $14$	<u>N</u> HBoc MeO <sub>2</sub> C 14a <sup>a</sup>	10	80
4	$\frac{NH_2.HC1}{HS}$	$HS \underbrace{\overset{NHBoc}{{\leftarrow}}}_{\text{CO}_2\text{Me}}$	10	81
5	$HN \bigvee N NH_2.HCl$ 16	RN N NHBoc <b>16b<sup>a</sup></b> R = H; <b>16b<sup>a</sup></b> R = Boc	10 15	82 80
6		Meo	25	88
7	MeO NH2 O NH2 O O NHOMe	MeO NHBoc	15	93
8	$18$ $MeO \xrightarrow{\text{NH}_2} 0 \text{ MH}_2$ $MeO \xrightarrow{\text{NH}_2} 0 \text{ MO}$	18a <sup>b</sup> MeO NHBoc MeO ''''O''''O'''''O'''''O'''''O'''''''''	20	89
	19	<b>19a</b> <sup>b</sup>		

<sup>a</sup> Commercially available compounds.

<sup>b</sup> All new compounds gave satisfactory spectral and analytical data.

#### Acknowledgement

P.S. and J.J.R. are thankful to the CSIR, New Delhi, India for financial support.

### **References and notes**

- (a) Greene, T. W.; Wuts, P. G. M. Protective Groups in Organic Synthesis, 2nd ed.; Wiley: New York, 1999; pp 503–550, and references cited therein; (b) Caroino, L. A. 1973, 6, 191; (c) Xiuo, X. Yi; Ngu, K.; Choa, C.; Patel, D. V. J. Org. Chem. 1997, 62, 6968.
- (a) Merrifield, R. B. J. Am. Chem. Soc. 1963, 85, 2149; (b) Merrifield, R. B. J. Am. Chem. Soc. 1964, 86, 304.
- Pandey, R. K.; Dagade, S. P.; Upadhyay, R. K.; Dongare, M. K.; Pradeep, K. Arkivoc 2002, VII, 28.
- Sharma, G. V. M.; Goverdhan Reddy, Ch.; Radha Krishna, P. J. Org. Chem. 2003, 68, 4574.
- (a) Sharma, G. V. M.; Goverdhan Reddy, V.; Subhash Chander, A.; Ravinder Reddy, K. *Tetrahedron: Asymmetry* 2002, 13, 21; (b) Sharma, G. V. M.; Ravinder Reddy, K.; Radha Krishna, P.; Ravi Shankar, A.; Narsimulu, K.; Kiran Kumar, S.; Jaya Prakash, P.; Jagannadh, B.; Kunwar, A. C. J. Am. Chem. Soc. 2003, 125, 13670.